

Selective Aptamers for Detection of Estradiol and Ethynylestradiol in Natural Waters

Spurti U. Akki,[†] Charles J. Werth,^{†,‡,*} and Scott K. Silverman^{†,‡,*}

[†]Department of Civil and Environmental Engineering, University of Illinois at Urbana-Champaign,
205 North Mathews Avenue, Urbana, Illinois 61801, United States

[‡]Department of Civil, Architectural and Environmental Engineering, University of Texas at Austin,
301 East Dean Keeton Street, Austin, Texas 78712, United States

[§]Department of Chemistry, University of Illinois at Urbana-Champaign,
600 South Mathews Avenue, Urbana, Illinois 61801, United States

Table of Contents

Supplemental Experimental Section	page S2
Selection progression	page S5
Sequences of individual aptamers	page S5
Equilibrium filtration assay data	page S6
Errors for equilibrium filtration assay data	page S8
DMS probing data	page S9
Lake and tap water characteristics.....	page S11
Relationship between K_d and LOD.....	page S12
References for Supporting Information.....	page S13

Contents: 13 pages with 7 figures and 2 tables

Supplemental Experimental Section

Materials. The DNA oligonucleotides (random pools) used for the E2 and EE aptamer selections were, respectively, 5'-CGAAGCGCTAGAACAT-N₄₀-AGTACATGAGACTTAGCTGATCCTGATGG-3' and 5'-CGAA-GTCGCCATCTCTTC-N₄₀-ATAGTGAGTCGTATTAAGCTGATCCTGATGG-3', where N₄₀ denotes 40 consecutive random nucleotides. In the PCR step of each selection, the primers were E2 forward 5'-CGAAGCGCTAGAACAT-3', E2 reverse 5'-(AAC)₄XCCATCAGGATCAGCTAAGTCTCATGTACT-3', EE forward 5'-CGAAGTCGCCATCTCTTC-3', and EE reverse 5'-(AAC)₄XCCATCAGGATCAGCT-3', where X = HEG spacer to stop Taq polymerase (thereby allowing separation of the two DNA single strands). For cloning the E2 selection, the forward and reverse primers were respectively modified to 5'-TAATTAATTAATTACGAAGCG-CTAGAACAT-3' and 5'-TAATTAATTAATTACCCATCAGGATCAGCT-3', each containing stop codons in all three reading frames. For cloning of the EE selection, only the reverse primer was modified by inclusion of stop codons.

Immobilization of E2 and EE on Agarose Support. For immobilization, approximately 25 mL of distilled water was added to 6 g of epoxy-activated agarose support in a 50 mL plastic (polyethylene) tube until the combined volume of swelled support and water was 33 mL (~15–20 min). The suspension was transferred to a 60 mL medium-porosity sintered glass funnel and washed by suction with 1200 mL of water in 30 mL portions with manual agitation over 3 h. The support (~15 mL volume) was transferred to a 50 mL plastic tube and washed with 3 × 15 mL of immobilization buffer [100 mM Na₂CO₃, 50% (v/v) isopropyl alcohol, pH 13], centrifuging at 250 × g for 5 min after each addition. The immobilization reaction was performed in a 15 mL plastic tube using 5 mL of support and 5 mL of immobilization buffer containing 20 mM E2 or EE for 8 h (E2) or 26 h (EE) at 37 °C; the EE immobilization was found empirically to require the longer time. The sample was centrifuged as above, and the immobilization buffer was decanted. The support was washed with 3 × 5 mL of immobilization buffer. The unreacted epoxy groups were capped by incubation with 5 mL of 1 M ethanolamine for 20 h at 23 °C, followed by centrifugation, decanting, and washing with three complete cycles of 5 mL of 0.1 M sodium acetate, pH 4.0, 5 mL of 0.1 M Tris, pH 8.0, and 5 mL of water. For preparation of the pre-selection support with only immobilized ethanolamine, the support was directly derivatized via the capping procedure. The stock of derivatized support was stored in 5 mL of 20% (v/v) ethanol at 4 °C.

To obtain a portion of derivatized support, the stock sample was vortexed, and a 400 µL aliquot of the suspension was removed to a 1.7 mL tube. The sample was centrifuged at 2000 × g for 10 s and the supernatant was removed by pipet, providing 200 µL of derivatized support.

The extent of derivatization of the support was quantified by UV-visible spectroscopy, by measuring the absorbance at 280 nm (A₂₈₀) of 75 µL of the support suspended in 225 µL of 50% (w/v) poly(ethylene glycol) (PEG 4000, Fluka). A calibration plot was constructed by measuring the A₂₈₀ values of a series of suspensions of 75 µL of unmodified support (Sephacrose 6B, Sigma, catalog number 6B100) in 222 µL of 50% (w/v) PEG 4000 containing 60–300 nmol of aqueous phase E2 or EE, along with 3 µL of ethanol for solubility. The calibration plot indicated that typically 50–100 nmol of E2 or EE was immobilized on 75 µL of support, equivalent to 133–267 nmol of E2 or EE on 200 µL of support, which was the amount used in each selection round.

In vitro Selection Procedure. The in vitro selection procedure was performed as follows.

Procedure for Initiating Selection (Round 1). The selection strategy for DNA aptamers is shown in Figure 1B. In the first selection round, 50 µL of pre-selection support was transferred to a Micro Bio-Spin chromatography column (Bio-Rad) and washed with 5 × 250 µL of binding buffer (50 mM Tris, pH 7.5, 5 mM MgCl₂, and 300 mM NaCl), each time centrifuging at 3000 × g for 30 s. To suppress nonspecific binding of DNA to the support, a “blocking” DNA oligonucleotide, (AAC)₂₀, was used. Approximately 100 pmol of the blocking oligonucleotide in 400 µL of binding buffer was added to the washed support in the pre-selection column. The suspension was transferred to a 2 mL tube sealed with an O-ring and incubated on a nutator at 23 °C for 10 min. The suspension was returned to the pre-selection column, which was centrifuged, and the filtrate was discarded. In a separate 1.7 mL tube, 500 pmol of a random DNA pool (~3 × 10¹⁴ unique sequences, of which 5 pmol was 5'-³²P-radiolabeled to enable monitoring of

binding activity) in 400 μL of binding buffer was annealed by heating at 95 $^{\circ}\text{C}$ for 5 min and cooling at 23 $^{\circ}\text{C}$ for 30 min. The annealed DNA pool was added to the washed support containing the blocking oligonucleotide. The suspension was transferred to the 2 mL O-ring tube, incubated on the nutator for 10 min, returned to the pre-selection column, and centrifuged. The pre-selection support was suspended in 400 μL of binding buffer, and the suspension was transferred to a 7 mL scintillation vial (sample A). The flow-through from the pre-selection column was retained for incubation with the selection support.

Into a separate column was placed 200 μL of E2/EE-derivatized selection support, which was washed with $10 \times 500 \mu\text{L}$ of binding buffer, centrifuging each time. The support was treated with 400 pmol of (AAC)₂₀ blocking oligonucleotide as described above. The retained flow-through from the pre-selection column was added to the blocking oligonucleotide-treated selection support. This suspension was transferred to a 2 mL O-ring tube, incubated on the nutator for 1 h, returned to the selection column, and centrifuged. The flow-through was transferred to a scintillation vial (sample B), and the selection support was retained for subsequent washes.

The selection support was washed with $10 \times 200 \mu\text{L}$ of wash buffer, which was the binding buffer containing 20% (v/v) ethanol for E2 support or 2% (v/v) ethanol for EE support. For each wash, the buffer was incubated in the column for 3 min followed by centrifugation. Two consecutive washes were combined and transferred to a scintillation vial (sample C). The selection support was eluted twice with free E2/EE in solution. For both elutions, 300 μL of the appropriate wash buffer containing 200 μM E2 or 20 μM EE was transferred to the selection column. The suspension was transferred to a 2 mL O-ring tube, incubated on the nutator for 2 h, returned to the selection column, and centrifuged. The two elutions were combined and transferred to a scintillation vial (sample D). A suspension of the E2/EE support in 400 μL of binding buffer was transferred to a scintillation vial (sample E).

A scintillation counter (Beckman Coulter LS 6500) was used to quantify the amount of ³²P in each of the five samples A–E. To track the selection progression (Figure S1), the fraction of total counts (samples A–E) that bound to the selection column and subsequently were eluted specifically with free E2/EE (sample D only) was used as quantification of the pool binding activity.

After counting, sample D was divided between two 1.7 mL tubes and precipitated with ethanol. Two PCR reactions were performed, i.e., 10-cycle PCR followed by 30-cycle PCR. First, a 100 μL sample was prepared containing the ethanol-precipitated product, 200 pmol of forward primer, 50 pmol of reverse primer, 20 nmol of each dNTP, and 10 μL of 10 \times Taq polymerase buffer (1 \times = 20 mM Tris-HCl, pH 8.8, 10 mM (NH₄)₂SO₄, 10 mM KCl, 2 mM MgSO₄, and 0.1% Triton X-100). This sample was cycled 10 times according to the following PCR program: 94 $^{\circ}\text{C}$ for 2 min, 10 \times (94 $^{\circ}\text{C}$ for 30 s, 47 $^{\circ}\text{C}$ for 30 s, 72 $^{\circ}\text{C}$ for 30 s), 72 $^{\circ}\text{C}$ for 5 min. Taq polymerase was removed by phenol/chloroform extraction. Second, a 50 μL sample was prepared containing 1 μL of the 10-cycle PCR product, 100 pmol of forward primer, 25 pmol of reverse primer, 10 nmol of each dNTP, 20 μCi of α -³²P-dCTP (800 Ci/mmol), and 5 μL of 10 \times Taq polymerase buffer. This sample was cycled 30 times according to the following PCR program: 94 $^{\circ}\text{C}$ for 2 min, 30 \times (94 $^{\circ}\text{C}$ for 30 s, 47 $^{\circ}\text{C}$ for 30 s, 72 $^{\circ}\text{C}$ for 30 s), 72 $^{\circ}\text{C}$ for 5 min. The sample was separated by 8% PAGE, extracted with TEN buffer, and precipitated with ethanol.

Procedure for Subsequent Selection Rounds 2+. Each subsequent selection round was performed as described for Round 1, with the following modifications. The pre-selection support was incubated simultaneously with annealed 30-cycle PCR product (~5-10 pmol) and 100 pmol of blocking oligonucleotide in 400 μL of binding buffer. 400 pmol of blocking oligonucleotide in 5 μL of binding buffer was added to the flow-through from the pre-selection column, before the sample was incubated with the selection support. After the selection support was washed with the appropriate wash buffer, the two elution steps were each performed for 30 min (1 h total). For rounds 9+ of the E2 selection, a longer elution time of 3 h for each elution step (6 h total) was used to promote elution of tighter-binding sequences.

Cloning and Sequencing. After 10 rounds for both the E2 and EE selections, the 30-cycle PCR was performed as described above, with the following modifications: 1 μL of 1/1000 dilution of the 10-cycle PCR product was used; α -³²P-dCTP was omitted; one or two cloning primers incorporating stop codons were used (see Materials); and 25 pmol of each cloning primer was used. The resulting 114 bp (from E2

selection) and 104 bp (from EE selection) double-stranded DNA product with single adenosine overhangs as added by Taq polymerase was isolated by 2% agarose gel (Fermentas gel extraction kit). Individual aptamers were cloned using a TOPO TA kit (Life Technologies). Miniprep DNA (Fermentas) was prepared, and the presence of the expected aptamer insert was confirmed by EcoRI digestion and 2% agarose gel electrophoresis. Sequencing was performed at the UIUC Biotechnology Center. Aptamers for preliminary binding studies were prepared by 30-cycle PCR directly from the miniprep DNA. For subsequent studies, aptamers were prepared by solid-phase synthesis (IDT). Aptamer sequences that were studied in detail are shown in Figure S2.

Preliminary Assays for Binding of Aptamers to Immobilized E2/EE. As a preliminary screen of individual clones, binding assays were performed. The pre-selection step was omitted; the E2/EE-derivatized support was incubated with an individual clone (obtained by 30 cycles of PCR using the miniprep DNA as template) followed by washing with the appropriate wash buffer as described above in the selection procedure. The column was eluted with 300 μ L portions of appropriate wash buffer containing increasing concentrations of E2/EE, ranging from 0.1 to 300 μ M. Each elution was collected separately for scintillation counting. Individual clones were characterized further when they showed substantial binding activity, as assessed by an increase in the amount of material eluted with increasing concentration of E2/EE.

Equilibrium Filtration Assay. The rationale for arriving at equation (4) in the main text is described here. Because of the molecular weight cutoff membrane, the retentate contains both A•L and free L, whereas the filtrate has only free L. The two experimentally measurable concentrations of L are total L in the retentate and free L in the filtrate, noting that there is no bound L in the filtrate (which has no aptamer). Moreover, the concentration of free L is the same in the retentate and filtrate, because the EDC passes freely through the cutoff filter. The scintillation counting data were correlated to molar concentrations by $(c_R + c_F)/(v_R + v_F) = c$ ($= 0.5 \mu$ M). From the experimental values of c_F/v_F , the concentration of free L was calculated and used to determine the concentration of A•L in the retentate. Finally, noting that the concentration, y , of A•L in the full sample is defined in reference to the total volume of the full sample ($v_R + v_F$), the value of y was calculated and plotted according to equation (3), as shown in Figure 2A.

DMS Probing. Each aptamer was probed using dimethyl sulfate (DMS) in the presence of varying concentrations of E2 or EE. In a 1.7 mL tube, 0.4 pmol of 5'-³²P-radiolabeled aptamer in 10 μ L of modified binding buffer (50 mM HEPES, pH 7.5, 5 mM MgCl₂, and 300 mM NaCl, where HEPES replaces Tris in the original binding buffer because Tris can be methylated by DMS) was annealed by heating at 95 °C for 5 min followed by cooling at 23 °C for 1.5 h. To this sample was added 10 μ L of modified binding buffer containing 5% (v/v) of ethanol and a particular concentration of E2/EE (final concentration 1 nM to 100 μ M). As a standard, a sample of 0.4 pmol of 5'-³²P-radiolabeled aptamer in 20 μ L of TE (10 mM Tris, pH 8.0, and 1 mM EDTA) without E2/EE was prepared and annealed. To 15 μ L of the aptamer sample was added 10 μ L of methylation buffer [50 mM sodium cacodylate, pH 8.0, 300 mM NaCl, 2.43% (v/v) ethanol, 1.25 % (v/v) DMSO, and 0.075% (v/v) Triton X-100] containing 0.27% (v.v) DMS. The 25 μ L sample was incubated at 23 °C for 30 min and precipitated with ethanol. The product was dissolved in 50 μ L of 10% (v/v) piperidine and heated at 95 °C for 30 min followed by cooling on dry ice for 5 min. The samples were dried under vacuum overnight and separated by 12% PAGE, with imaging by PhosphorImager and quantification by ImageQuant (GE Healthcare).

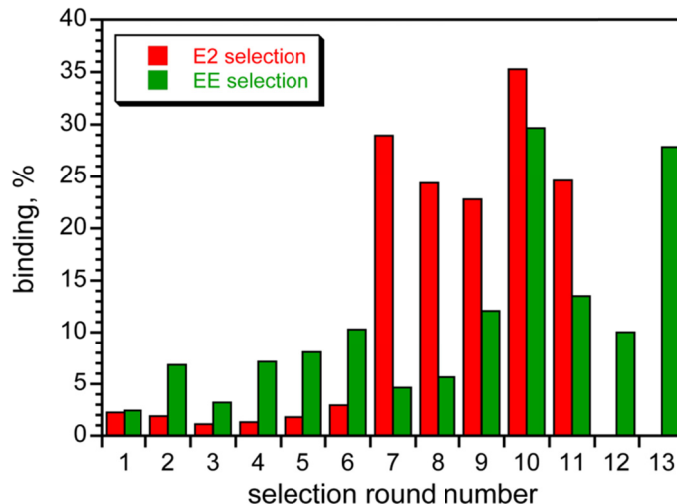
Selection progressions

Figure S1. Selection progressions for E2 and EE selections. Long elution time pressure was applied from Round 9 for E2 selections only. Individual aptamers were cloned from the Round 10 pools in both selections once sustained stable activity was observed.

Sequences of individual deoxyribozymes

		10		20		30		40											
E2Apt1	GG	C	CGGGG	A	GG	CAGGGG	-A	GAGTGACACG	CGGTCGGTGA	C	40								
E2Apt2	T	.AC	.	GG	.ATCTG	.GA	.AC	.TG	.	.	.	ATCCGTG	T	41	
EEApt1	A	.	CATC	.AC	.	.	TC	.AAA	-G	CTAAC	.T	.	.A	.	.CAAA	.C	.G	T	40
EEApt2	A	.	GGAAC	.T	.	.	GGA	.CT	-T	.C	.G	.CATAC	AAACTATG	.G	A			40	

Figure S2. Sequences of E2 and EE aptamers that were characterized by equilibrium filtration and DMS probing assays. Only the initially random (N_{40}) sequences are shown. All aptamers were used as 5'-CGAAGCGCTAGAACAT- N_{40} -AGTACATGAGACTTAGCTGATCCTGATGG-3' (E2Apt1, E2Apt2) and 5'-CGAAGTCGCCATCTCTTC- N_{40} -ATAGTGAGTC-GTATTAAGCTGATCCTGATGG-3' (EEApt1, EEApt2). Identical residues in comparison to E2Apt1 are represented as dots. Gaps are depicted as dashes. The E2Apt2 sequence includes one additional nucleotide (41 nt), apparently due to a spontaneous insertion by Taq polymerase during an unknown selection round.

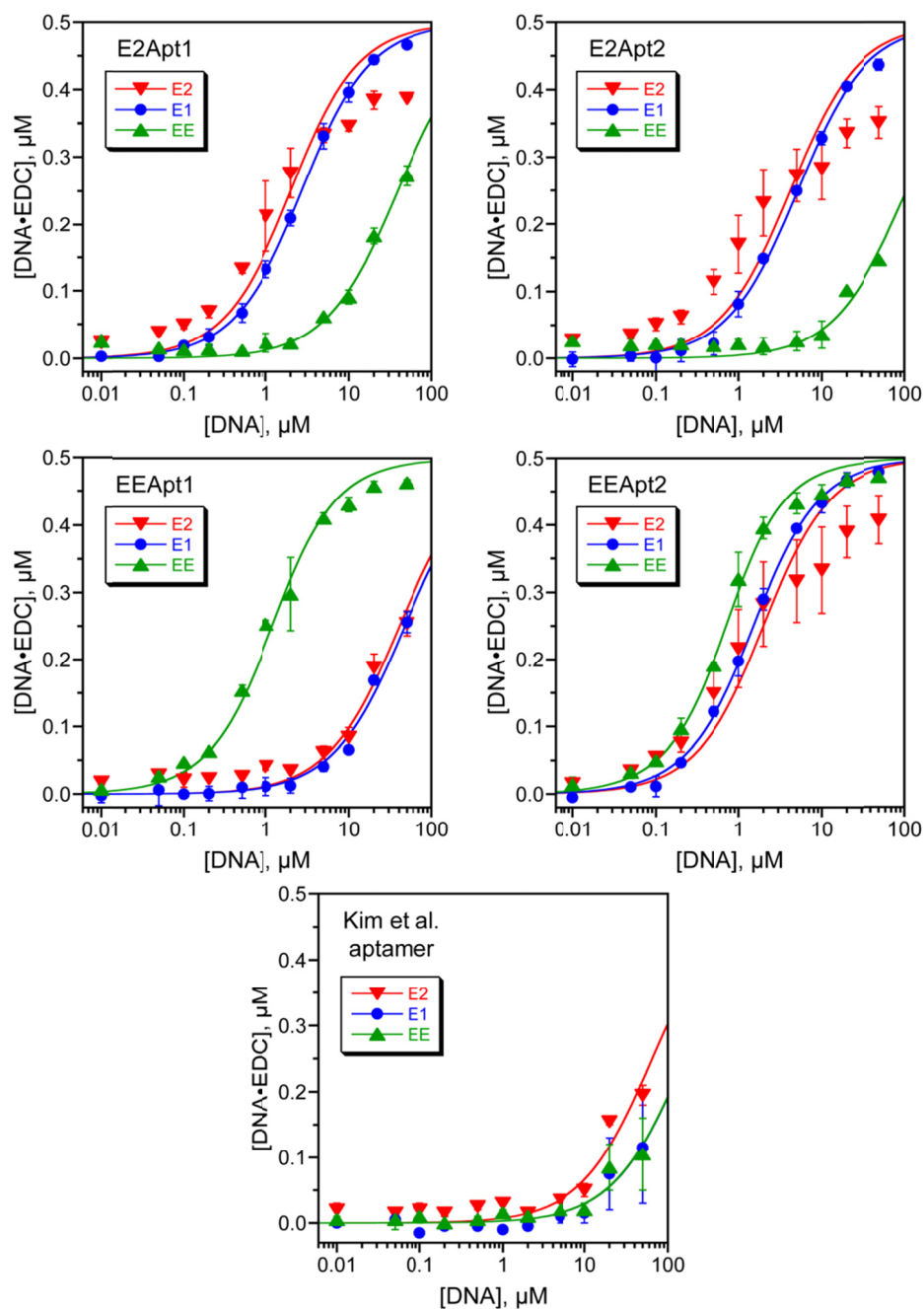
Equilibrium filtration assay data

Figure S3. Equilibrium filtration assay data for the E2Apt1, E2Apt2, EEApt1, and EEApt2 aptamers as well as the E2 aptamer identified by Kim et al.¹ These data are fit using the model that assumes Hill coefficient $n = 1$. K_d values are tabulated in Table 1. See Figure S4 for alternate fitting approach with Hill coefficient $n \neq 1$ for the E2 data.

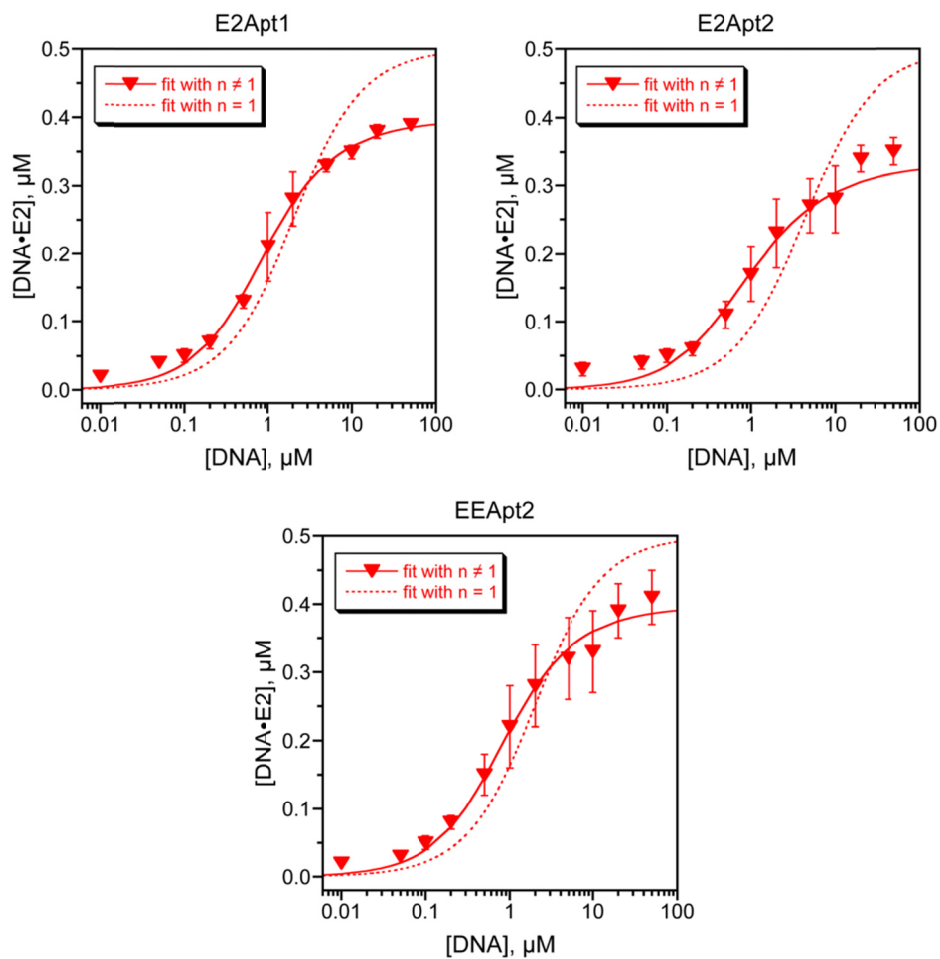


Figure S4. Equilibrium filtration assay data for the E2Apt1, E2Apt2 and EEAp22 aptamers binding to E2, fit using the model with Hill coefficient $n \neq 1$. K_d and n values are tabulated in Table 1. Fits to the same data with $n = 1$ from Figure S3 are shown for comparison. There is no plot for EEAp21 because this aptamer did not bind appreciably to E2.

Errors for equilibrium filtration assay parameters

Best-fit parameter values (K_d , n) were obtained using equation (7) and minimizing the root mean square error (RMSE) given by

$$RMSE = \sqrt{\frac{\sum (y_{\text{expt}} - y_{\text{pred}})^2}{N}}$$

where N is the number of data points ($N = 11$), y_{expt} is the experimentally obtained value of $A \cdot L$, and y_{pred} is the value of $A \cdot L$ computed using specific values of K_d and n . In order to obtain error bars (standard errors) for the best-fit values of parameters, K_d and n , we first calculated the covariance matrix V . V is computed using variance (σ^2) and the first derivative matrix X for each data point, where

$$V = (X^T \cdot X)^{-1} \sigma^2$$

$$\sigma^2 = \frac{RMSE^2 \times N}{N - p}$$

$$X_{i1} = \frac{\partial y_{i,\text{pred}}}{\partial K_d} = \frac{y_{\text{pred}}(K_d + \Delta K_d) - y_{\text{pred}}(K_d)}{\Delta K_d}$$

$$X_{i2} = \frac{\partial y_{i,\text{pred}}}{\partial n} = \frac{y_{\text{pred}}(n + \Delta n) - y_{\text{pred}}(n)}{\Delta n}$$

where p is the number of parameters ($p = 2$ in our case), and the parameters are increased by 15% to compute the first derivatives, e.g., $\Delta K_d = 0.15 \cdot K_d$. The standard error for K_d is given by $\sqrt{V_{11}}$ and that for n is given by $\sqrt{V_{22}}$.

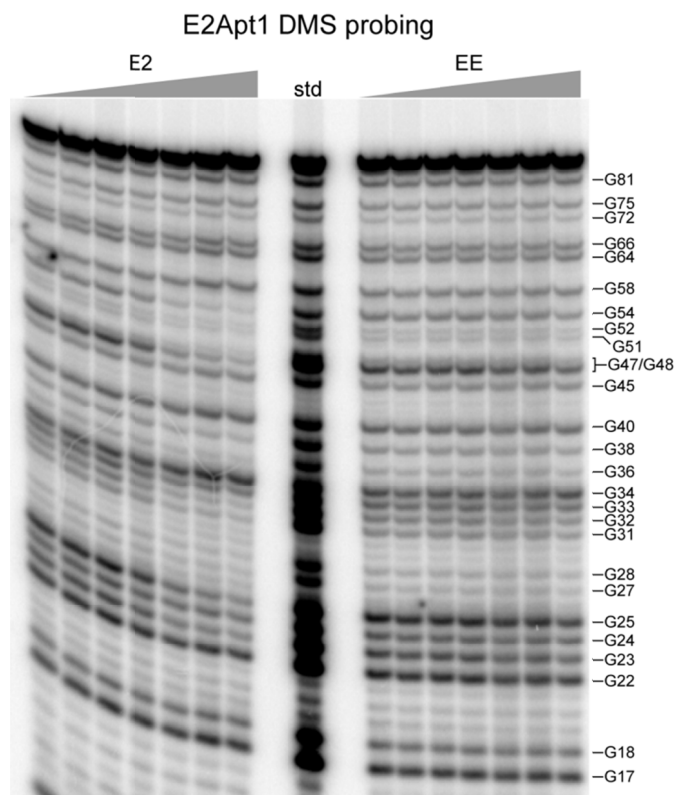
DMS probing data

Figure S5. Full PAGE image for the experiment shown in Figure 4B. For each set of lanes, E2/EE concentrations are (left to right) 0, 0.001, 0.01, 0.1, 1, 10, and 100 μM .

Table S1. K_d values determined from DMS probing data for individual guanosine nucleotides of E2Apt1, E2Apt2, and EEApt1 with E2, E2, and EE, respectively. Values were derived from data such as that shown in Figure 4C. Error bars are from the fits to equation (6) as described in the Experimental Section. For nucleotide G64 of EEApt1, the normalized band intensity was observed to increase two-fold as the EE concentration was increased; this was the only nucleotide in any of these aptamers for which the normalized band intensity increased rather than decreased.

E2Apt1 nt	E2 K_d , μM	E2Apt2 nt	E2 K_d , μM	EEApt1 nt	EE K_d , μM
G54	0.24 ± 0.08	G54	0.65 ± 0.21	G64	1.5 ± 0.5
G47/G48	0.23 ± 0.07	G48	0.75 ± 0.35	G62	3.7 ± 1.0
G32	0.17 ± 0.04	G46	0.77 ± 0.14	G37	4.0 ± 1.9
G31	0.47 ± 0.13	G45	0.80 ± 0.10	G30	1.3 ± 0.5
G25	0.31 ± 0.07	G31	0.52 ± 0.12		
		G30	0.80 ± 0.16		
		G29	0.65 ± 0.15		
		G28	0.70 ± 0.19		
		G27	0.78 ± 0.15		
		G24	0.63 ± 0.28		

Lake and tap water characteristics**Table S2.** Characteristics of the lake and tap waters used in the equilibrium filtration assay. The water samples were characterized by the Illinois State Water Survey lab facility in Champaign. nd = not detected.

Component	Units	Lake	Tap
pH		8.29	8.47
alkalinity	mg/L as CaCO ₃	141	135
B	mg/L	0.11	0.36
Ba	mg/L	0.06	0.08
Ca	mg/L	47.4	13.8
Cu	mg/L	nd	0.08
K	mg/L	3.40	2.58
Mg	mg/L	15.9	12.5
Mn	mg/L	0.003	nd
Na	mg/L	113	32
S	mg/L	5.38	nd
Si	mg/L	0.99	3.63
Sr	mg/L	0.10	0.15
Tl	mg/L	0.02	nd
F	mg/L	0.18	0.86
Cl	mg/L	182.1	13.2
NO ₃ (as N)	mg/L	nd	0.05
SO ₄	mg/L	14.69	0.32
TDS	mg/L	477	166
dissolved TOC	mg/L	4.43	1.25

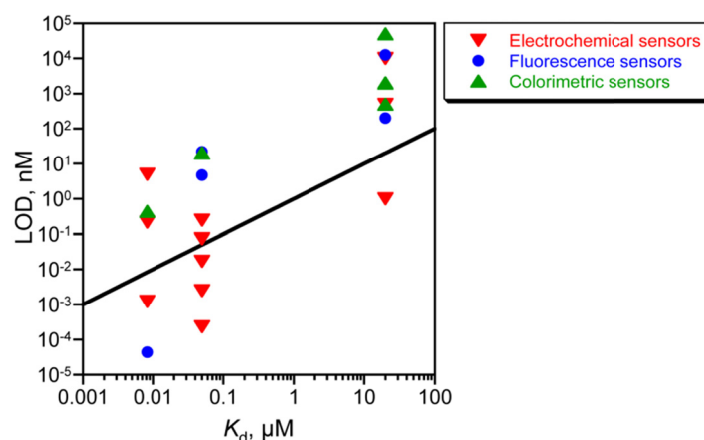
Relationship between K_d and LOD

Figure S7. LOD versus K_d for three broad categories of platforms: electrochemical sensors, fluorescence sensors, and colorimetric sensors. The cited studies involve aptasensors developed for three small molecules: bisphenol A ($K_d = 8.3$ nM),²⁻⁶ ochratoxin A ($K_d = 50$ nM),⁷⁻¹³ and cocaine ($K_d = 20$ μM).¹⁴⁻²² The slope of the straight line is 1.

References for Supporting Information

- (1) Kim, Y. S.; Jung, H. S.; Matsuura, T.; Lee, H. Y.; Kawai, T.; Gu, M. B. Electrochemical detection of 17 β -estradiol using DNA aptamer immobilized gold electrode chip. *Biosens. Bioelectron.* **2007**, *22*, 2525-2531.
- (2) Xue, F.; Wu, J.; Chu, H.; Mei, Z.; Ye, Y.; Liu, J.; Zhang, R.; Peng, C.; Zheng, L.; Chen, W. Electrochemical aptasensor for the determination of bisphenol A in drinking water. *Microchim. Acta* **2013**, *180*, 109-115.
- (3) Ragavan, K. V.; Selvakumar, L. S.; Thakur, M. S. Functionalized aptamers as nano-bioprobes for ultrasensitive detection of bisphenol-A. *Chem. Commun.* **2013**, *49*, 5960-5962.
- (4) Mei, Z.; Chu, H.; Chen, W.; Xue, F.; Liu, J.; Xu, H.; Zhang, R.; Zheng, L. Ultrasensitive one-step rapid visual detection of bisphenol A in water samples by label-free aptasensor. *Biosens. Bioelectron.* **2013**, *39*, 26-30.
- (5) Zhou, L.; Wang, J.; Li, D.; Li, Y. An electrochemical aptasensor based on gold nanoparticles dotted graphene modified glassy carbon electrode for label-free detection of bisphenol A in milk samples. *Food Chem.* **2014**, *162*, 34-40.
- (6) Chen, L.; Zeng, X.; Ferhan, A. R.; Chi, Y.; Kim, D.-H.; Chen, G. Signal-on electrochemiluminescent aptasensors based on target controlled permeable films. *Chem. Commun.* **2015**, *51*, 1035-1038.
- (7) Wang, Z.; Duan, N.; Hun, X.; Wu, S. Electrochemiluminescent aptamer biosensor for the determination of ochratoxin A at a gold-nanoparticles-modified gold electrode using *N*-(aminobutyl)-*N*-ethylisoluminol as a luminescent label. *Anal. Bioanal. Chem.* **2010**, *398*, 2125-2132.
- (8) Kuang, H.; Chen, W.; Xu, D.; Xu, L.; Zhu, Y.; Liu, L.; Chu, H.; Peng, C.; Xu, C.; Zhu, S. Fabricated aptamer-based electrochemical "signal-off" sensor of ochratoxin A. *Biosens. Bioelectron.* **2010**, *26*, 710-716.
- (9) Yang, C.; Wang, Y.; Marty, J.-L.; Yang, X. Aptamer-based colorimetric biosensing of Ochratoxin A using unmodified gold nanoparticles indicator. *Biosens. Bioelectron.* **2011**, *26*, 2724-2727.
- (10) Sheng, L.; Ren, J.; Miao, Y.; Wang, J.; Wang, E. PVP-coated graphene oxide for selective determination of ochratoxin A via quenching fluorescence of free aptamer. *Biosens. Bioelectron.* **2011**, *26*, 3494-3499.

- (11) Prabhakar, N.; Matharu, Z.; Malhotra, B. D. Polyaniline Langmuir-Blodgett film based aptasensor for ochratoxin A detection. *Biosens. Bioelectron.* **2011**, *26*, 4006-4011.
- (12) Tong, P.; Zhang, L.; Xu, J.-J.; Chen, H.-Y. Simply amplified electrochemical aptasensor of ochratoxin A based on exonuclease-catalyzed target recycling. *Biosens. Bioelectron.* **2011**, *29*, 97-101.
- (13) Wu, J.; Chu, H.; Mei, Z.; Deng, Y.; Xue, F.; Zheng, L.; Chen, W. Ultrasensitive one-step rapid detection of ochratoxin A by the folding-based electrochemical aptasensor. *Anal. Chim. Acta* **2012**, *753*, 27-31.
- (14) Stojanovic, M. N.; de Prada, P.; Landry, D. W. Aptamer-based folding fluorescent sensor for cocaine. *J. Am. Chem. Soc.* **2001**, *123*, 4928-4931.
- (15) Stojanovic, M. N.; Landry, D. W. Aptamer-based colorimetric probe for cocaine. *J. Am. Chem. Soc.* **2002**, *124*, 9678-9679.
- (16) Baker, B. R.; Lai, R. Y.; Wood, M. S.; Doctor, E. H.; Heeger, A. J.; Plaxco, K. W. An electronic, aptamer-based small-molecule sensor for the rapid, label-free detection of cocaine in adulterated samples and biological fluids. *J. Am. Chem. Soc.* **2006**, *128*, 3138-3139.
- (17) Liu, J.; Lu, Y. Fast Colorimetric Sensing of Adenosine and Cocaine Based on a General Sensor Design Involving Aptamers and Nanoparticles. *Angew. Chem. Int. Ed.* **2006**, *45*, 90-94.
- (18) Li, Y.; Qi, H.; Peng, Y.; Yang, J.; Zhang, C. Electrogenerated chemiluminescence aptamer-based biosensor for the determination of cocaine. *Electrochem. Commun.* **2007**, *9*, 2571-2575.
- (19) Li, X.; Qi, H.; Shen, L.; Gao, Q.; Zhang, C. Electrochemical Aptasensor for the Determination of Cocaine Incorporating Gold Nanoparticles Modification. *Electroanalysis* **2008**, *20*, 1475-1482.
- (20) Zhang, J.; Wang, L.; Pan, D.; Song, S.; Boey, F. Y. C.; Zhang, H.; Fan, C. Visual cocaine detection with gold nanoparticles and rationally engineered aptamer structures. *Small* **2008**, *4*, 1196-1200.
- (21) Jiang, B.; Wang, M.; Chen, Y.; Xie, J.; Xiang, Y. Highly sensitive electrochemical detection of cocaine on graphene/AuNP modified electrode via catalytic redox-recycling amplification. *Biosens. Bioelectron.* **2012**, *32*, 305-308.
- (22) Roncancio, D.; Yu, H.; Xu, X.; Wu, S.; Liu, R.; Debord, J.; Lou, X.; Xiao, Y. A label-free aptamer-fluorophore assembly for rapid and specific detection of cocaine in biofluids. *Anal. Chem.* **2014**, *86*, 11100-11106.