





Cite this: *Org. Biomol. Chem.*, 2021, **19**, 171

DNAzymes for amine and peptide lysine acylation†

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DNAzymes were previously identified by *in vitro* selection for a variety of chemical reactions, including several biologically relevant peptide modifications. However, finding DNAzymes for peptide lysine acylation is a substantial challenge. By using suitably reactive aryl ester acyl donors as the electrophiles, here we used *in vitro* selection to identify DNAzymes that acylate amines, including lysine side chains of DNA-anchored peptides. Some of the DNAzymes can transfer a small glutaryl group to an amino group. These results expand the scope of DNAzyme catalysis and suggest the future broader applicability of DNAzymes for sequence-selective lysine acylation of peptide and protein substrates.

Received 2nd October 2020,
Accepted 28th October 2020

DOI: 10.1039/d0ob02015j

rsc.li/obc

Introduction

DNAzymes, also called DNA enzymes and deoxyribozymes, are specific DNA sequences that catalyze chemical reactions, similar to protein enzymes as catalytic amino acid sequences.^{1–8} Nature evolved both protein enzymes and RNA enzymes (ribozymes), but to date all DNAzymes have been identified in the laboratory by *in vitro* selection.^{9–13} The earliest-reported and most-studied DNAzymes catalyze RNA cleavage by transesterification at phosphorus.^{1–4,14–18} Since then, DNAzymes for a range of reactions and substrates have been found. Our laboratory is interested in DNAzymes that catalyze reactions relevant to protein post-translational modifications (PTMs),^{19,20} such as phosphorylation,^{21–23} dephosphorylation,²⁴ and amide hydrolysis.^{25,26} One such reaction is lysine (Lys) acylation, where Lys acetylation is critical for histones and in other contexts,^{27–30} and many longer-chain Lys acylation PTMs^{31–33} such as malonylation,^{34,35} succinylation,^{34,36} and glutarylation^{37,38} have been discovered yet are poorly understood.³⁹ As an alternative to approaches that include introduction of Lys analogues,^{40–43} nonsense codon suppression,^{44–49} bottom-up ligation-based assembly strategies,^{50–52} or enzymatic methods that typically require creation of a non-native protein by insertion of a specific target sequence,^{53–58} DNAzymes are promising for top-down introduction of Lys acylation PTMs onto intact native proteins,^{59–65} but only if DNAzymes can be identified with the fundamental catalytic ability of Lys acylation.

Toward this goal, we previously reported the first DNAzymes that catalyze Lys modification of any kind.⁶⁶ We used 5'-phosphorimidazolide (5'-Imp) DNA as the electrophile, resulting in the formation of a Lys-phosphoramidite bond (Fig. 1A). From that study, a key lesson was the need for a suitably reactive electrophile to react with the amine nucleophile, a consideration that outweighed the value of highly preorganizing the two substrates. Therefore, to achieve Lys acylation in our present work, we carefully considered our options for the acyl donor electrophile. Arguably the most straightforward choice is a thioester, considering that nature often uses thioe-

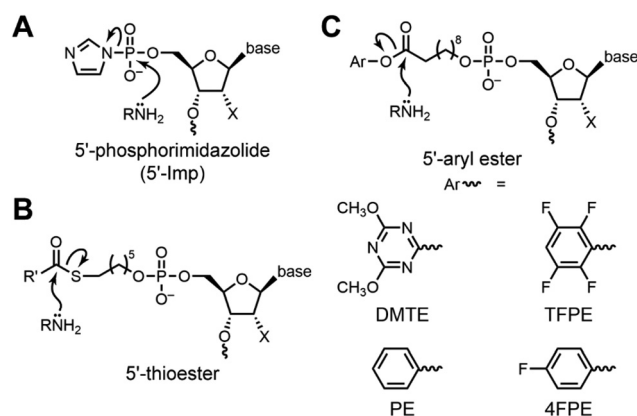


Fig. 1 Electrophilic reaction partners for amine nucleophiles in DNAzyme-catalyzed reactions. (A) 5'-Phosphorimidazolide (5'-Imp) DNA, for which we previously found DNAzymes that catalyze Lys-phosphoramidite formation.⁶⁶ (B) 5'-Thioester DNA, for which here we were unable to identify any amine-acylating DNAzymes. (C) 5'-Aryl ester DNA, for which here we describe new DNAzymes that catalyze amine acylation, including with Lys peptides. The DMTE, TFPE, PE, and 4FPE aryl ester substrates were evaluated during this study. DMTE = 4,6-dimethoxy-1,3,5-triazin-2-yl ester; TFPE = 2,3,5,6-tetrafluorophenyl ester; PE = phenyl ester; 4FPE = 4-fluorophenyl ester.

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† Electronic supplementary information (ESI) available. See DOI: 10.1039/d0ob02015j

sters as acyl donors, including for amine acylation (Fig. 1B). However, in several new *in vitro* selection efforts, some of which included modified DNA nucleotides that previously led to amide-hydrolyzing DNAzymes,²⁶ we were unable to identify any amine-acylating DNAzymes using thioesters. We therefore turned to aryl esters (Fig. 1C) as acyl donors, among other considerations noting their tunable electrophilicity. Here we report the outcome of these *in vitro* selection experiments, culminating in new DNAzymes that catalyze Lys acylation of DNA-anchored peptide substrates.

Results and discussion

Thioester acyl donor electrophile

We first performed a set of *in vitro* selection experiments with an amine nucleophile and a thioester acyl donor electrophile, where both substrates were conjugated to DNA anchor oligonucleotides for standard Watson-Crick binding to the initially random DNAzyme pool (Fig. S1†). Each experiment used either the four canonical DNA nucleotides, or dT was replaced with one of several modified nucleotides as we reported for DNAzyme-catalyzed amide hydrolysis.²⁶ However, in all cases, after 10 selection rounds no amine acylation activity was observed. We concluded that a thioester is insufficiently reactive as an electrophile to allow the identification of amine-acylating DNAzymes, and a more reactive acyl donor is required.

Highly reactive aryl ester acyl donor electrophiles

We then performed *in vitro* selection experiments with acyl donor oligonucleotides activated *in situ* from their 5'-carboxylic acid (5'-CO₂H) precursors using two common amide-forming

coupling reagents, 4-(4,6-dimethoxy-1,3,5-triazin-2-yl)-4-methylmorpholinium chloride (DMT-MM)⁶⁷ or the combination of the water-soluble 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide (EDC) and 2,3,5,6-tetrafluorophenol (TFP).⁶⁸⁻⁷¹ Before performing *in vitro* selection, the resulting DMT and TFP esters (here abbreviated DMTE and TFPE; structures in Fig. 1C) were assayed for their uncatalyzed background reactivities, using a DNA splint complementary to the DNA-anchored acyl donor and the simple DNA-anchored amine nucleophile (DNA-C₃-NH₂; Fig. 2A and B). Both DMTE and TFPE led to relatively high uncatalyzed background reactivity, with substantial formation of acylation product; e.g., 34% amide formation in 0.5 min (DMTE) and 15% amide formation in 0.5 min (TFPE), each assessed at pH 7.0. Nevertheless, *in vitro* selection still had the potential to lead to DNAzymes with rate enhancement above this uncatalyzed background reaction. We therefore proceeded to perform *in vitro* selection experiments using the DMTE and TFPE acyl donors with the DNA-C₃-NH₂ substrate (Fig. 2C).

These *in vitro* selection experiments used N₄₀ initially random regions (where N₄₀ is a compromise between longer random regions that may be able to form more complex structures, and shorter random regions for which more of sequence space is explored⁷²) along with incubation conditions identical to those used for the background assays. For *in vitro* selection using the DMTE acyl donor, these incubation conditions included 50 mM DMT-MM, 100 mM MOPS, pH 7.0, 1 mM ZnCl₂, and 150 mM NaCl at 37 °C for 2 h. For the TFPE acyl donor, these conditions included 50 mM each EDC and TFP, 100 mM MOPS, pH 7.0, 40 mM MgCl₂, 20 mM MnCl₂, 1 mM ZnCl₂, and 150 mM NaCl at 37 °C for 2 h. For both selection experiments using the DNA-C₃-NH₂ substrate, the pool yield

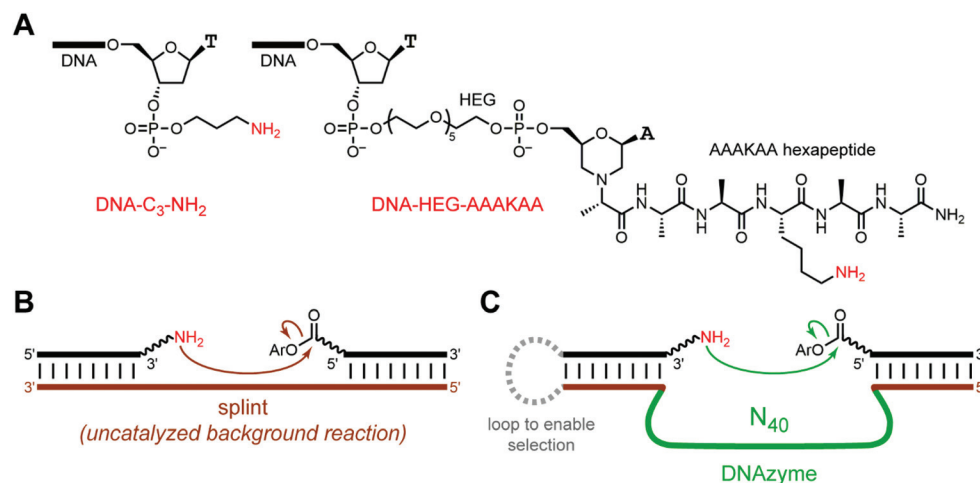


Fig. 2 *In vitro* selection for identifying amine-acylating DNAzymes. (A) Structures of the amine nucleophiles, DNA-C₃-NH₂ and DNA-HEG-AAAKAA. The latter is formed by reductive amination with the free α -NH₂ of the AAAKAA hexapeptide. The HEG-hexapeptide tether is longer than the C₃ tether by 35 atoms. (B) Uncatalyzed, splinted background reaction between DNA-anchored amine and aryl ester. (C) Key step of *in vitro* selection. See Fig. S2 and ESI† text for details. Any DNAzyme sequences that catalyze amine acylation by the 5'-aryl ester oligonucleotide grow larger by the size of that oligonucleotide, which allows PAGE-shift separation of those DNAzyme sequences. The DNA population is therefore enriched in catalytically active sequences in each round. Iteration for multiple selection rounds is required because some noncatalytic sequences can also survive any particular selection round. The connecting loop on the left side is not included when individual DNAzymes are assayed. The linker joining the aryl ester and the 5'-end of the oligonucleotide is (CH₂)₉ and a 5'-phosphate.

increased during the selection rounds to 37% (DMTE) or 13% (TFPE) at round 4 (Fig. S3A†), at which point individual DNA sequences were identified by cloning and sequencing (Fig. S4A†). For each selection, however, the emergent DNA sequences had no rate enhancement above the uncatalyzed, splinted background reaction under the same incubation conditions. We concluded that each individual DNA sequence likely adopts a combination of secondary and tertiary structure that merely recapitulates a complementary splint. Apparently, rate enhancement beyond the splinting effect cannot be achieved because the DMTE or TFPE electrophile is too reactive.

The same outcome of finding DNA sequences that have no rate enhancement was found for the DMTE and TFPE acyl donors when the amine substrate was instead a hexa(ethylene glycol) [HEG]-tethered AAKAA hexapeptide that included a single Lys residue (Fig. 2A, S3B, and S4B†). Therefore, using the considerably less preorganized and presumably less reactive DNA-HEG-AAKAA substrate did not suppress the too-high background reactivity of the DMTE and TFPE acyl donors.

Intermediate-reactivity aryl ester acyl donor electrophiles and simple amine nucleophile (DNA-C₃-NH₂)

To this point in our efforts, the observed selection outcomes were divergent. With a thioester acyl donor, no DNAszymes were found due to the insufficiently reactive electrophile. In contrast, with DMTE or TFPE as the acyl donors, specific DNA sequences emerged from the selection process, but they lacked rate enhancement beyond a splint because these electrophiles were too reactive. Therefore, we turned our attention to acyl

donors with intermediate reactivity. TFP forms an aryl ester, the parent compound of which is the simple phenyl ester, and the DMT ester formed upon activation of a carboxylic acid by DMT-MM is a multiply substituted aryl ester. We therefore investigated the phenyl ester (PE) and 4-fluorophenyl ester (4FPE) acyl donor substrates (Fig. 1C). For these experiments, we decided to synthesize and purify each new 5'-aryl ester oligonucleotide substrate rather than rely upon *in situ* activation, as we did with the reagents DMT-MM and EDC/TFP because of the high reactivity of the corresponding DMT and TFP esters. The PE and 4FPE oligonucleotide substrates were synthesized from the 5'-CO₂H oligonucleotide, EDC, and the appropriate phenol derivative, followed by HPLC purification (Fig. S5†).

With the PE and 4FPE 5'-aryl ester oligonucleotide substrates in hand, we first evaluated their hydrolytic stabilities and uncatalyzed, splinted background reactivities with the DNA-C₃-NH₂ substrate under likely incubation conditions for *in vitro* selection, with quantitative details provided in the ESI (Table S1†). Based on the data, we chose two particular incubation conditions for each acyl donor substrate: lower pH of 7.5 (70 mM HEPES, pH 7.5, 40 mM MgCl₂, 20 mM MnCl₂, 1 mM ZnCl₂, and 150 mM NaCl at 37 °C for 16 h) and higher pH of 9.0 (50 mM CHES, pH 9.0, 40 mM MgCl₂, and 150 mM NaCl at 37 °C for 16 h). Each of Mg²⁺, Mn²⁺, and Zn²⁺ was included at pH 7.5 on the basis of our many prior successful DNAszyme selection efforts using these metal ions at pH 7.5. However, at pH 9.0 only Mg²⁺ can be included, because Mn²⁺ oxidizes and Zn²⁺ precipitates at this higher pH. With the

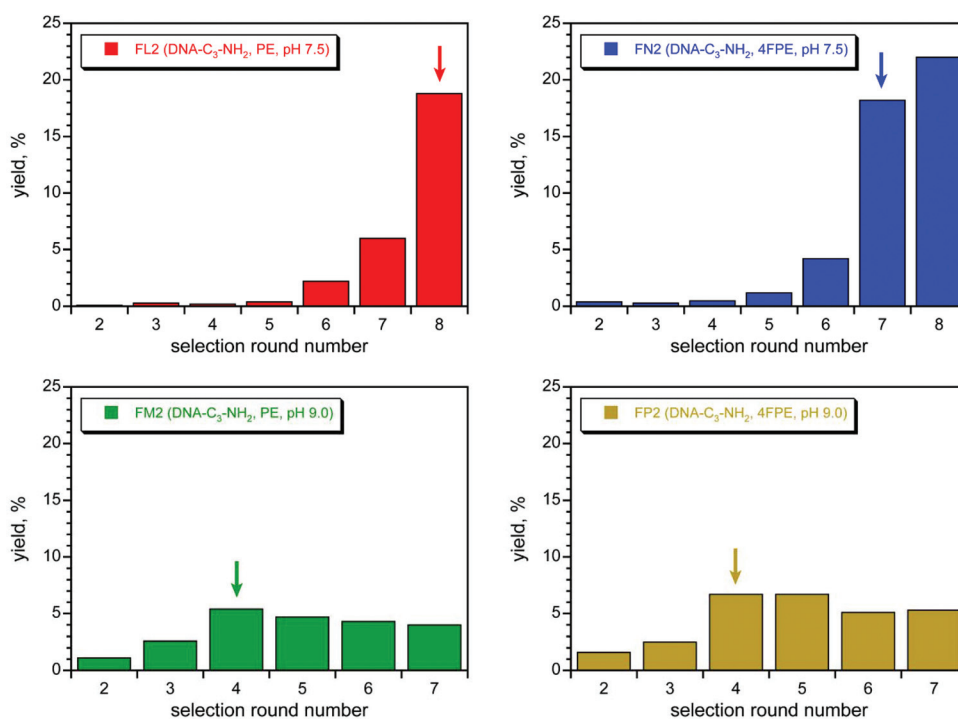


Fig. 3 Selection progressions using the DNA-C₃-NH₂ nucleophile with the PE and 4FPE acyl donors. See text for details of incubation conditions. Arrows mark the cloned rounds. In all cases, the round 1 yield was not quantified because nonradiolabeled pool was used.

more reactive 4FPE substrate at the higher pH of 9.0, the acyl donor oligonucleotide was still 74% intact after 16 h, with only 3.5% splinted background yield (Table S1†). All other substrate and pH combinations had even higher intact acyl donor and even lower background yield.

With two acyl donor oligonucleotide substrates (PE and 4FPE) and two incubation conditions (pH 7.5 and 9.0) per substrate, the four selection experiments were each iterated for 7–8 rounds (Fig. 3). For each substrate, the pH 7.5 selection gave a substantial and promising increase in pool yield as the rounds progressed; these selections were cloned after round 8 (PE) and 7 (4FPE). Substantial amine acylation activity was observed for many of the resulting individual DNAzymes (Fig. 4A and S6;† sequences in Fig. S4C†). The PE and 4FPE selections each gave four DNAzymes, each with ~50% yield in 24 h. The highest rate enhancements, calculated by taking single-turnover k_{obs} for the DNAzyme and dividing by k_{bkgd} for the uncatalyzed, splinted background reaction using the complementary DNA splint, were 1100 and 760 (each $\sim 10^3$) for the

7FN216 and 7FN202 DNAzymes, respectively (Fig. 4B), which both use the 4FPE substrate at pH 7.5.⁷³

In contrast, the pH 9.0 selection for each substrate gave only a modest increase in pool yield as the rounds progressed (Fig. 3). Each selection was cloned after round 4 and gave six (PE) or seven (4FPE) DNAzymes, with only 5–20% yields. Each individual DNAzyme had low rate enhancement of at most 2 (representative data in Fig. 4C; sequences in Fig. S4D†), and these DNAzymes were not studied further. An immediate conclusion is that the two pH 7.5 selections, for which the incubation conditions led to lower background yields (0.3–0.6%; Table S1†), were more successful at providing DNAzymes than the two pH 9.0 selections, which had higher background yields (2.4–3.5%).

We investigated the metal ion dependence of the eight DNAzymes that use DNA-C₃-NH₂ and the PE or 4FPE substrates at pH 7.5. By evaluating each DNAzyme with all possible combinations of 40 mM Mg²⁺, 20 mM Mn²⁺, and 1 mM Zn²⁺, which were the concentrations of each ion that were used

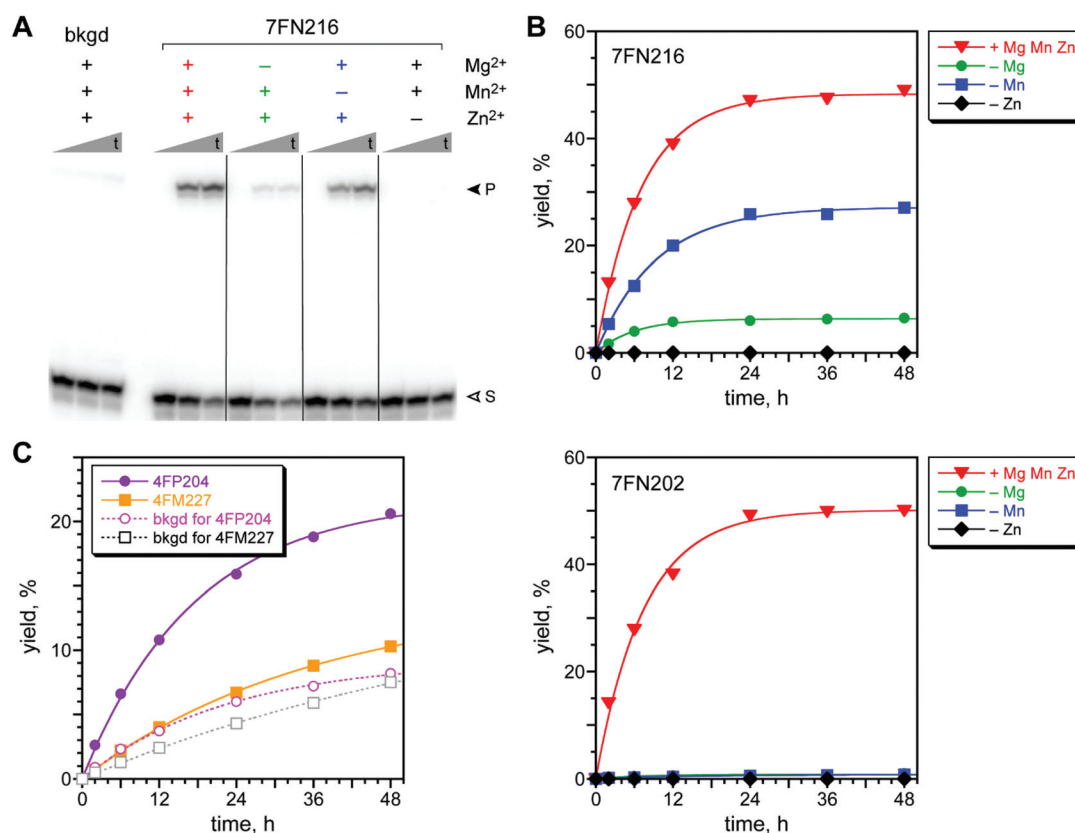


Fig. 4 Assays of DNAzymes identified by *in vitro* selection using the DNA-C₃-NH₂ substrate. (A) Representative PAGE image for the 7FN216 DNAzyme with its 4FPE acyl donor substrate. Bkgd = complementary DNA splint in place of DNAzyme, to assess the uncatalyzed background reaction. Incubation conditions: 70 mM HEPES, pH 7.5, combinations of 40 mM MgCl₂, 20 mM MnCl₂, and 1 mM ZnCl₂ as indicated, and 150 mM NaCl at 37 °C. The background reaction was with Mg²⁺/Mn²⁺/Zn²⁺. Shown are representative timepoints ($t = 0.5$ min, 6 h, 48 h; S = substrate, P = product). (B) Kinetic plots for 7FN216 and 7FN202, which have different metal ion dependence. Plots for the other two 7FN2 DNAzymes (4FPE substrate) and all four 8FL2 DNAzymes (PE substrate) are in Fig. S6.† k_{obs} values ($\text{h}^{-1} \pm$ standard deviation, each $n = 3$, with Mg²⁺/Mn²⁺/Zn²⁺): 7FN216, 0.19 ± 0.06 ; 7FN202, 0.13 ± 0.02 ; background (k_{bkgd}), $(1.7 \pm 0.1) \times 10^{-4}$. k_{obs} values for the other six DNAzymes are 0.05–0.09 h^{-1} . (C) Kinetic plots for 4FM227 (PE substrate), 4FP204 (4FPE substrate), and a complementary DNA splint as background reaction for each substrate. Incubation conditions: 50 mM CHES, pH 9.0, 40 mM MgCl₂, and 150 mM NaCl at 37 °C. k_{obs} values (h^{-1} , $n = 1$): 4FM227, 0.028; 4FP204, 0.057; background (k_{bkgd}): PE, 0.014; 4FPE, 0.044. Data was similar for the other eleven 4FM2 and 4FP2 DNAzymes (not shown).

during the selection process, we found two types of metal ion dependence (Fig. 4). All eight DNAszymes worked optimally when all three of Mg^{2+} , Mn^{2+} , and Zn^{2+} were included. Five of the eight DNAszymes, such as 7FN216, retained substantial activity with only Mg^{2+} and Zn^{2+} (omitting Mn^{2+}) and had greatly reduced yield with only Mn^{2+} and Zn^{2+} (omitting Mg^{2+}), whereas omitting Zn^{2+} led to no activity. The other three DNAszymes, such as 7FN202, needed all three of Mg^{2+} , Mn^{2+} , and Zn^{2+} for catalysis.

For one representative DNAszyme from each of the pH 7.5 selection experiments, the acylation product was isolated by PAGE, and its expected mass was confirmed by MALDI mass spectrometry (see ESI† for numerical data). In addition, all eight of the pH 7.5 DNAszymes were assayed using, as a negative control in place of DNA- C_3-NH_2 , an unmodified DNA oligonucleotide lacking the pendant C_3-NH_2 at its 3'-end. In each case, no product formation was observed (<0.2%; data not shown), consistent with nucleophilic reactivity of the amino group in the DNAszyme-catalyzed acylation reaction.

Intermediate-reactivity aryl ester acyl donors and peptide lysine nucleophile (DNA-HEG-AAAKAA)

With success in identifying DNAszymes that acylate the simple amine nucleophile DNA- C_3-NH_2 , especially at the lower pH of 7.5, we shifted our attention to the substrate that presents a peptide Lys nucleophile, DNA-HEG-AAAKAA, where our ultimate

goal is DNAszyme-catalyzed acylation of peptide and protein Lys residues. Unsurprisingly based on our previous report with DNA-catalyzed Lys phosphoramidate formation,⁶⁶ none of the above-described DNAszymes identified using the DNA- C_3-NH_2 substrate had any detectable activity (<0.2%) when tested with DNA-HEG-AAAKAA (data not shown). Therefore, we performed new selection experiments with DNA-HEG-AAAKAA. We used the same pair of selection conditions as for DNA- C_3-NH_2 , each with the same two PE and 4FPE acyl donor substrates, and the four selections were iterated for 11 rounds (Fig. 5). For three of the selections, those with the PE substrate at pH 7.5 and 9.0 as well as that with the 4FPE substrate at pH 9.0, substantial Lys acylation activity was found, and all three selections were cloned after round 11. The fourth selection with the 4FPE substrate at pH 7.5 was not cloned because of poor pool activity (see Fig. 5 caption).

Lys acylation activity was observed for many of the resulting individual DNAszymes (Fig. 6). The selection experiment at pH 7.5 with the PE substrate gave one single DNAszyme sequence, 11HB201 (Fig. S4E†), with modest 15% yield consistent with that of the uncloned round 11 pool as a whole. In contrast, the two pH 9.0 selections both led to several distinct DNAszymes (Fig. S4F†), most of which have high yields (>75%). The rate enhancements at pH 9.0 with Mg^{2+} for 11HC206 (PE substrate) and 11HF210 (4FPE substrate) were 86 and 60, respectively. These values are about an order of magnitude lower than the

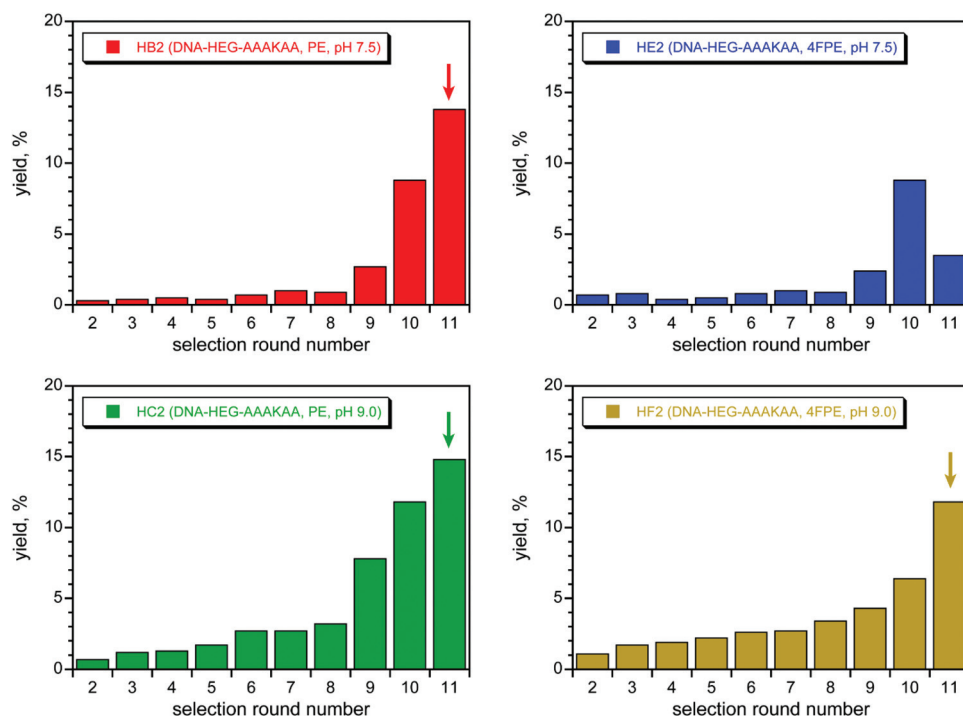


Fig. 5 Selection progressions using the DNA-HEG-AAAKAA nucleophile with the PE and 4FPE acyl donors. See text for details of incubation conditions. Arrows mark the cloned rounds. In all cases, the round 1 yield was not quantified because nonradiolabeled pool was used. The HE2 (pH 7.5, 4FPE) selection was not cloned because multiple bands were observed in the product region of the gel, and most of these were assigned to non-catalytic DNA sequences that misfold and therefore migrate aberrantly. Consistent with this decision, each of the 11HB2, 11HC2, and 11HF2 pools was active in *trans*, i.e., with the DNA pool not ligated to the DNA-HEG-AAAKAA substrate, while in contrast, the 10HE2 pool was inactive in *trans*.

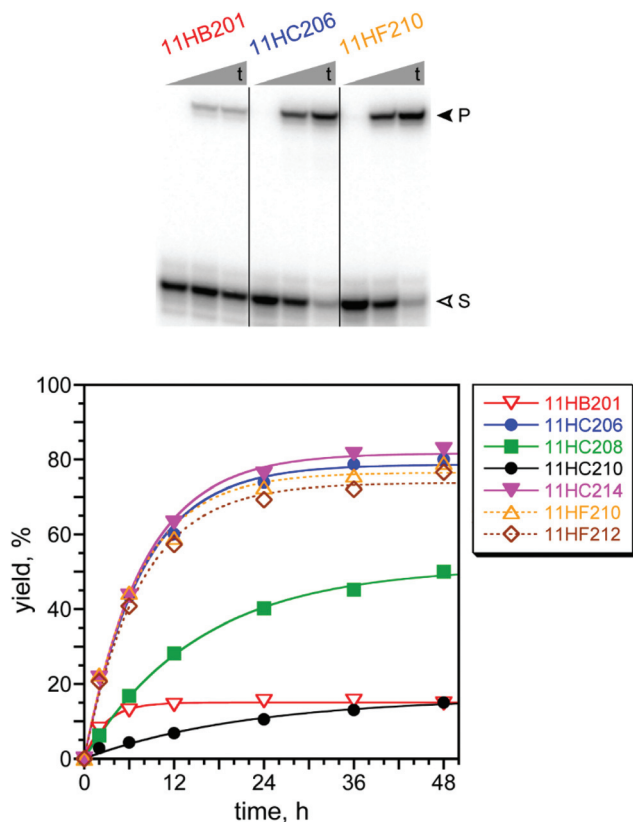


Fig. 6 Assays of DNAzymes identified by *in vitro* selection using the DNA-HEG-AAAKAA substrate with the PE and 4FPE acyl donors. Incubation conditions for 11HB201 (PE substrate): 70 mM HEPES, pH 7.5, 40 mM MgCl₂, 20 mM MnCl₂, 1 mM ZnCl₂, and 150 mM NaCl at 37 °C. Incubation conditions for all 11HC2 (PE substrate) and 11HF2 (4FPE substrate) DNAzymes: 50 mM CHES, pH 9.0, 40 mM MgCl₂, and 150 mM NaCl at 37 °C. The PAGE assays for three DNAzymes are shown with representative timepoints (*t* = 0.5 min, 6 h, 48 h; *S* = substrate, *P* = product). Representative kinetic plots are shown for several DNAzymes. k_{obs} values ($\text{h}^{-1} \pm$ standard deviation, $n = 3$): 11HB201, 0.27 ± 0.08 ; 11HC206, 0.12 ± 0.02 ; 11HF210, 0.12 ± 0.02 . Additional k_{obs} values (h^{-1} with % yield at 48 h, $n = 1$): 11HC208, 0.065 (50%); 11HC210, 0.046 (15%); 11HC214, 0.13 (83%); 11HF212, 0.13 (77%). Background assays used the DNA-HEG-AAAKAA substrate and an exactly complementary splint in place of a DNAzyme, with k_{bkgd} values (h^{-1}) and % yield at 48 h as follows: pH 7.5 PE, 0.00013 ± 0.00004 (0.6%; $n = 3$); pH 9.0 PE, 0.0014 ± 0.0001 (5.8%; $n = 3$); pH 9.0 4FPE, 0.0020 ± 0.0001 (8.0%; $n = 4$).

rate enhancements for the best DNAzymes identified for acylation of the DNA-C₃-NH₂ substrate, such as 7FN216 and 7FN202. This is due to the different pH values, 9.0 for the DNA-HEG-AAAKAA DNAzymes *versus* 7.5 for the DNA-C₃-NH₂ DNAzymes, where k_{bkgd} is substantially greater at the higher pH, and a greater k_{bkgd} leads to a lower rate enhancement. MALDI mass spectrometry was consistent with Lys acylation for the three DNAzymes of Fig. 6 (see ESI†). Interestingly, in an outcome opposite to that of the above-described selections with the simpler DNA-C₃-NH₂ substrate, here the selections with DNA-HEG-AAAKAA at the higher pH of 9.0 were more successful in leading to active DNAzymes.

Control experiments were performed using DNA-C₃-NH₂ in place of DNA-HEG-AAAKAA for all seven of the new DNAzymes. The results with the three DNAzymes of Fig. 6 are representative and also consistent with our previous report with DNA-catalyzed Lys phosphoramidate formation.⁶⁶ In 24 h, the yields with DNA-C₃-NH₂ were 11HB201, <0.1%; 11HC206, 0.8%; and 11HF210, 1.4%. The corresponding k_{obs} values are calculated to be >6400-fold, 360-fold, and 210-fold lower, respectively than k_{obs} of the same DNAzymes with DNA-HEG-AAAKAA (Fig. 6), which supports the conclusion of selective DNAzyme-catalyzed nucleophilic reactivity of the Lys amino group of the AAKAA hexapeptide. Because primary amino groups are not indiscriminately acylated by these DNAzymes, productive catalytic interactions are likely between each DNAzyme and its tethered peptide substrate.

Assays with free peptide substrates

In the longer term, peptide-modifying DNAzymes will have their greatest utility when they can function with free (untethered, not DNA-anchored) peptide substrates. We assayed the DNAzymes that were identified with the DNA-HEG-AAAKAA substrate for their ability to function with 2 mM of free AAKAA hexapeptide that is not tethered to the DNA anchor oligonucleotide. Unfortunately, in all cases, no activity was observed (<1.5% by PAGE-shift analysis, using 3'-³²P-radio-labeled 5'-aryl ester oligonucleotide; the untethered synthetic precursor oligonucleotide that was formerly connected to AAKAA was included in these experiments). This result is unsurprising, given that the DNA-anchored AAKAA was presented to the DNAzyme population in every selection round, and an analogous tether requirement by emergent DNAzymes has been encountered in many of our prior selections. In the future, we intend to perform lysine acylation selection experiments in which an azide-modified free peptide is used in the selection step, thereby enforcing a strict pressure for the resulting DNAzymes to function with the free peptide.⁷⁴

Assays with acyl donor for amine glutarylation

By the design of Fig. 2C, a successful DNAzyme-catalyzed acylation reaction joins the acyl donor oligonucleotide to the amine acceptor, and the small-molecule phenol derivative of the 5'-aryl ester oligonucleotide is the leaving group. Ideally, amine-acylating DNAzymes will instead be able to use an acyl donor that transfers a small-molecule acyl group rather than a large acyl-oligonucleotide. To explore this possibility, we synthesized a glutaryl donor oligonucleotide (Fig. 7A) in which the orientation of the aryl ester functional group was inverted, such that acylation results in glutarylation of the amine nucleophile. This was achieved by first preparing a glutaryl-azide small-molecule compound, which was then used in a CuAAC (copper-catalyzed azide-alkyne cycloaddition) reaction with a 5'-alkyne-modified oligonucleotide, to form the glutaryl donor oligonucleotide.

The glutaryl donor oligonucleotide is a *p*-alkyl-substituted aryl ester of glutaric acid, where the *p*-alkyl group is inherently electron-donating. Therefore, we expect the glutaryl donor

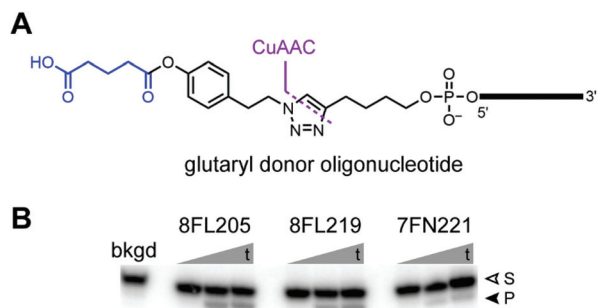


Fig. 7 DNAzyme assays with the glutaryl donor. (A) Glutaryl donor oligonucleotide structure, as prepared by CuAAC between a synthesized glutaryl-azide small-molecule compound and a 5'-alkyne-modified oligonucleotide. The glutaryl fragment (blue) is transferred during an amine acylation reaction. (B) Assay results using the glutaryl donor with three DNAzymes identified by *in vitro* selection for acylation of DNA-C₃-NH₂ with the PE (8FL205, 8FL219) or 4FPE (7FN221) acyl donor substrate. Bkgd = complementary DNA splint in place of DNAzyme, to assess the uncatalyzed background glutarylation reaction. Incubation conditions: 70 mM HEPES, pH 7.5, 40 mM MgCl₂, 20 mM MnCl₂, 1 mM ZnCl₂, and 150 mM NaCl at 37 °C. Shown are representative timepoints (t = 0.5 min, 16 h, 48 h; S = substrate, P = product). The respective yields at 48 h were 19%, 14%, and 5.4%, versus <0.8% for splinted background (no product band detectable). For each DNAzyme, no product was detectable with unmodified DNA in place of DNA-C₃-NH₂.

oligonucleotide to be somewhat less reactive than the PE substrate and perhaps substantially less reactive than the fluoro-substituted esters of Fig. 1C, including 4FPE. Also, the spatial presentations of the acyl donors are different (*e.g.*, the linkers to the oligonucleotide are not the same), which may suppress the ability of any DNAzymes identified by *in vitro* selection with the Fig. 1C acyl donors to function with the glutaryl donor oligonucleotide. Nevertheless, assaying the eight pH 7.5 DNAzymes of Fig. 4 with the glutaryl donor oligonucleotide revealed that three of these DNAzymes catalyze substantial glutarylation of the DNA-C₃-NH₂ substrate (Fig. 7B; see ESI† for mass spectrometry product confirmation). The glutarylation yield was as high as 19% for 8FL205, as compared to the splinted background yield of <0.8%. In contrast, none of the seven DNAzymes of Fig. 6 catalyzed observable glutarylation of DNA-HEG-AAKAA above the splinted background (data not shown). These findings establish the feasibility of DNAzyme-catalyzed amine glutarylation, and in ongoing work, we are performing new selection experiments aimed at directly identifying DNAzymes that catalyze Lys glutarylation.

Structural and mechanistic considerations

Using mfold,⁷⁵ we systematically predicted the secondary structures of all 28 of the new DNAzymes reported in this study. Each DNAzyme is predicted to have 1–7 plausible secondary structures, each with typically modest folding free energy in the range of –4 to –1 kcal mol^{–1}, although with three examples of folding free energies in the –9 to –5 kcal mol^{–1} range (Table S2 and Fig. S7†). In many cases, the various mfold-predicted secondary structures for a single DNAzyme are incompatible with one another. We have not endeavored to

synthesize and study the large number of DNAzyme mutants (including covariations) that would be required to assess experimentally the validity of the predicted secondary structures. Importantly, doing so would not provide much if any useful, actionable information for our future studies of amine acylation DNAzymes. In parallel, the three-dimensional structures of these new DNAzymes are unknown, in the context that only two DNAzyme structures of any kind have been reported.^{76,77} Comprehensive analysis of the secondary and tertiary structures and mechanisms of amine acylation DNAzymes will require efforts beyond the scope of the present study.

Conclusions

In this study we established experimentally that DNAzymes can catalyze amine acylation, including acylation of a Lys residue in a short DNA-anchored peptide. Key to this success was identifying suitably substituted aryl esters (PE and 4FPE) as the electrophilic acyl donors, along with appropriate incubation conditions that balance electrophile stability and reactivity. Thioesters were too unreactive to support DNAzyme catalysis, whereas the more electrophilic DMTE and TFPE were too reactive. The observation that different pH values (and therefore different degrees of uncatalyzed background reactivity) supported emergence of the best DNAzymes for the two different substrates, DNA-C₃-NH₂ and DNA-HEG-AAKAA, suggests that pH is an important experimental variable to explore in our future studies. The observation that several DNAzymes can function for amine glutarylation, even though these DNAzymes were not directly identified by selection for amine acylation using the glutaryl donor substrate, bodes well for the longer-term prospects of DNAzymes for amine and Lys acylation with biologically relevant small acyl groups. Finally, we anticipate that new selection experiments involving azide-modified peptides that are not anchored to a DNA oligonucleotide should enable identification of DNAzymes that function with free peptides, as we found for a tyrosine modification reaction.⁷⁴ Such DNAzymes may also be able to accept larger protein substrates for modification of surface-exposed side chains.

For several other DNAzyme-catalyzed activities, we previously found that performing *in vitro* selection using a peptide substrate with mixed amino acid composition led to DNAzymes that require those specific peptide sequences in their substrates.^{22,74,78} By analogy, we anticipate that for DNAzyme-catalyzed peptide Lys acylation, future selection experiments using mixed-composition Lys-containing peptides will provide sequence-selective Lys-acylating DNAzymes, including those that function with free peptide substrates when an appropriate selection pressure is imposed.⁷⁴ Expanding the substrate tolerance of such DNAzymes from peptides to proteins is a further challenge, but worth undertaking considering the difficulty inherent to achieving nonenzymatic site-selective Lys modification of native proteins.^{59–65} We are currently pursuing such experiments.

Experimental section

DNA oligonucleotides were obtained from Integrated DNA Technologies (Coralville, IA) or prepared by solid-phase synthesis on an ABI 394 instrument using reagents from Glen Research, including the 5'-CO₂H modifier (5'-carboxy-modifier C10, 10-1935). All oligonucleotides and conjugates except for 5'-CO₂H, 5'-aryl ester, and glutaryl donor oligonucleotides were purified by 7 M urea denaturing 20% or 8% PAGE with running buffer 1× TBE (89 mM each Tris and boric acid and 2 mM EDTA, pH 8.3), extracted from the polyacrylamide with TEN buffer (10 mM Tris, pH 8.0, 1 mM EDTA, 300 mM NaCl), and precipitated with ethanol. 5'-Aryl ester (PE and 4FPE) oligonucleotides were prepared from the HPLC-purified 5'-CO₂H oligonucleotide by treatment with EDC and phenol or 4-fluorophenol followed by HPLC purification; see ESI† for details. The glutaryl donor oligonucleotide was prepared by CuAAC using a 5'-alkyne oligonucleotide, as described in the ESI.† The AAKAA hexapeptide was prepared by solid-phase synthesis using Fmoc Rink amide MBHA resin as described.²⁴ The peptide was coupled to the DNA anchor oligonucleotide by reductive amination with a periodate-oxidized 3'-terminal rA nucleotide as described.²⁴ After the DNA-anchored hexapeptide was precipitated with ethanol, the Lys(Tfa) protecting group was removed by incubation in 30% aqueous NH₄OH at room temperature for 1 h, dried by SpeedVac, and purified by 20% PAGE. Procedures for selection, cloning, and initial analysis of individual clones are in the ESI.†

The general single-turnover assay procedure for each DNAzyme using a 5'-aryl ester oligonucleotide substrate was as follows. The DNA-anchored amine substrate was 5'-³²P-radiolabeled using γ-³²P-ATP and T4 polynucleotide kinase. A 14 μL sample containing 0.5 pmol of 5'-³²P radiolabeled amine substrate, 10 pmol of DNAzyme, and 20 pmol of 5'-aryl ester substrate was annealed in 5 mM HEPES, pH 7.5 or 5 mM CHES, pH 9.0, 15 mM NaCl, and 0.1 mM EDTA by heating at 95 °C for 3 min and cooling on ice 5 min. The DNAzyme-catalyzed reaction was initiated by bringing the sample to 20 μL total volume containing 70 mM HEPES, pH 7.5, 40 mM MgCl₂, 20 mM MnCl₂, 1 mM ZnCl₂ and 150 mM NaCl or 50 mM CHES, pH 9.0, 40 mM MgCl₂, and 150 mM NaCl. For the DMTE and TFPE substrates, the 5'-CO₂H substrate was used, and either DMT-MM or EDC/TFP was included for activation of the 5'-CO₂H group. The sample was incubated at 37 °C. At each time point, a 2 μL aliquot was quenched with 5 μL of stop solution (80% formamide, 1× TBE [89 mM each Tris and boric acid and 2 mM EDTA, pH 8.3], 50 mM EDTA, 0.025% bromophenol blue, 0.025% xylene cyanol). Before PAGE for most assays, to each quenched sample was added 80 pmol of a "decoy oligonucleotide", which was a 60-mer complementary to the DNAzyme's initially random region (40 nt) along with 10 nt of binding arm on either side. This decoy oligonucleotide was added to displace the DNAzyme from the substrate and product. In these cases when the decoy was omitted, gel bands were noticeably smeared, which inhibited proper quantification. Quenched samples were separated by 20% PAGE and

quantified using a Phosphorimager. Values of k_{obs} were obtained by fitting the yield *versus* time data directly to first-order kinetics; *i.e.*, $\text{yield} = Y \cdot (1 - e^{-kt})$, where $k = k_{\text{obs}}$ and Y is the final yield. Each k_{obs} value is reported with error calculated as the standard deviation from the indicated number of independent determinations. When k_{obs} was sufficiently low such that an exponential fit was not meaningful (*e.g.*, for the background reactions), the initial points were fit to a straight line, and k_{obs} was taken as the slope of the line.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The early part of this research was partially supported by NIH grant GM065966 to S. K. S.

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